

Solutions Chapter 2

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Solution: 2.33. 19.5g of CH₂FCOOH is dissolved in 500g of water. The depression in the freezing point of water observed is 1.0°C. Calculate the van's Hoff factor and dissociation constant of fluoroacetic acid. Solution: 2.34. Vapour pressure of water at 293 K is 17-535 mm Hg.

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